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Interlaboratory comparison study of two gas chromatography injection techniques viz. liquid and dynamic headspace for trace level quantification of Ethylene Oxide (EtO) and 2-Chloroethanol (2-CE) in ginger powder sample by using tandem mass spectrometry.

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1. Purpose

EtO is one of the most widely produced chemicals worldwide. It is colorless, odorless, Table 2: Methods with different experiment conditions flammable gaseous cyclic ether with boiling point of 10.4°C. It has a very strong antibacterial property. Due to its small size, it shows a high diffusivity and strong penetrating properties and is thus very effective in the disinfestation / disinfection of food commodities. EtO is highly carcinogenic, mutagenic and genotoxic impurity for living being. The US National Institute of Health (NIH) classified EtO as "known to be a human carcinogen based on sufficient evidence of carcinogenicity from studies in humans, including epidemiological studies and studies on mechanisms of carcinogenesis." Considering carcinogenicity and no acceptable threshold for exposure, no Acceptable Daily Intake (ADI) was established for EtO.

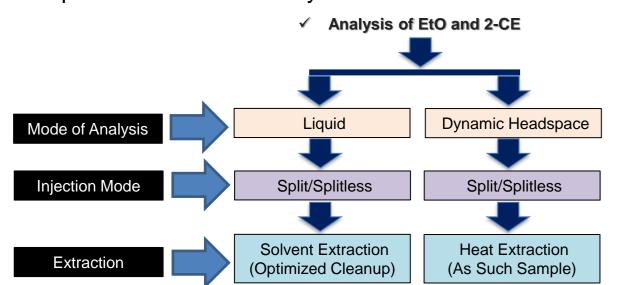
2-CE, which is one of the prominent metabolite of EtO is also considered to be genotoxic and potentially carcinogenic. Given the inconclusive toxicological picture of 2-CE, it was decided by regulatory authorities to follow the precaution approach and consider 2-CE equally toxic to EtO. Hence it is very important to quantitate EtO and 2-CE in food matrices at trace levels. EU-MRLs (Maximum Residue Levels as per European Commission) for EtO and 2-CE are summarized in Table 1

Table 1: EU-MRLs for EO and 2-CE

			Instrume
No.	Products	EU-MRLs for EtO and 2-CE	and 2-C
1	Tea, cocoa, spices	0.10 mg/kg	Table 3. In
2	Nuts, oil fruits, oilseeds	0.05 mg/kg	
3	Fruits, fungi, pulses	0.02 mg/kg	GCMS S Column
4	Cereals	0.02 mg/kg	Coldinii
5	Apicultural products	0.05 mg/kg	Injection

Various methods have been presented in EURL-SRM-Analytical Observation Report for the Flow Control analysis of EtO and 2-CE. The choice of the analytical approach depends on both the Carrier Gas analytes and the matrix to be analyzed. So, it is important to have a detailed comparison Column Flo study of different analytical approaches to ensure robustness, reliability and accuracy of the Linear Veloc method for the quantification of EtO and 2-CE.

This study covers the comparison between liquid injection technique and dynamic headspace injection technique using different extraction processes for ensuring highest extraction efficiencies. Incurred sample is used for this study.



In above study, Shimadzu's triple quadrupole GCMS-TQ8050 NX equipped with HS-20 NX dynamic headspace sampler and AOC-20i/s liquid sampler was used for analysis (Figure 1)



Exp No.	Liquid Method (Solvent Extraction)	Headspace Method (Heat Extraction)
1	Sample + Aluminium beads (Vortex for 15 min at 25°C)	Vial incubation at 65°C
2	Sample + Aluminium beads (Vortex for 45 min at 25°C)	Vial incubation at 125°C
3	Sample + Aluminium beads (Vortex for 90 min at 25°C)	Vial incubation at 65°C (Interlaboratory Study)
4	Sample + Aluminium beads (Sonicate for 90 min at 60°C)	

3. Methods

Instrument parameters are given in Table 3, whereas optimized MRM transitions of EtO CE are given in Table 4.

Instrument parameters for GC-MS/MS and HS

Syste

Temp. Progr

GC Run Tim Ionization M Interface Te Ion Source

<u>Headspace</u> Oven Temp. Sample Line Transfer Lin Trap Coolin Trap Desor Trap Equilib Shaking Lev Equilibrating Multi Injectio Pressurizing GC Cycle 7

Table 4: MRM

ID	Name
1	EtO
2	2-CE

44>29

80>31

Figure 1: Shimadzu GCMS-TQ8050 NX with AOC-20i+s and HS-20 NX

2. Design of Experiment

ip. MIC) sure for EtO and	: 150 °C : -10 °C : 280 °C : -10 °C : 5 : 25 mir : 10 : 75 kPa : 35 mir	; ; n a		
ıp. MIC)	: -10 °C : 280 °C : -10 °C : 5 : 25 mir : 10 : 75 kPa	; ; n a		
ıp. MIC)	: -10 °C : 280 °C : -10 °C : 5 : 25 mir : 10	; ; 1		
ıp.	: -10 °C : 280 °C : -10 °C : 5 : 25 mir			
	: -10 °C : 280 °C : -10 °C : 5			
	: -10 ºC : 280 ºC : -10 ºC	;)		
	: -10 ºC : 280 ºC	;)		
	: -10 ºC	;		
	: 150 °C	2		
_	: 65 °C	and 125 °C		
S				
: 24(O°C			
		ation (EI)		
: 20	min			
	30	235	8.33	
	-	35	5.0	
	•	(⁰ C)	(min)	_
		Temp.	Hold Tin	ne
•	: Column Flow			
: Spl	: Split (10:1) for liquid method			
	-	-	1.D., 1.4 μm α	
	-			r
	: SH (S/ : Sp : Sp : Co : He : 2.0 : 36. Ra (' : 20 : Ele : 250 : 240	: SH-502.2, 60 (S/N: 227-363 : Split (10:1) for : Split (20:1) for : Column Flow : Helium : 2.0 mL/min : 36.0 cm/s Ramp Rate ($^{\circ}$ C/min) : 20 min : Electron Ioniza : 250 $^{\circ}$ C : 240 $^{\circ}$ C	: SH-502.2, 60 m, 0.25 mm (S/N: 227-36341-03) : Split (10:1) for liquid meth : Split (20:1) for dynamic he : Column Flow : Helium : 2.0 mL/min : 36.0 cm/s Ramp Rate Temp. (°C/min) (°C) - 35 30 235 : 20 min : Electron Ionization (EI) : 250 °C : 240 °C	: Split (10:1) for liquid method : Split (20:1) for dynamic headspace meth : Column Flow : Helium : 2.0 mL/min : 36.0 cm/s Ramp Rate Temp. Hold Tin (°C/min) (°C) (min) - 35 5.0 30 235 8.33 : 20 min : Electron Ionization (EI) : 250 °C : 240 °C : 65 °C and 125 °C

44>28

80>44

4. Results and Discussion

Summary for linearity is given in Table 5, whereas experimental results are given in Table 6

Table 5: Summary for linearity

Details

Liquid - Matrix Matc

(100,200,400,600 and 8

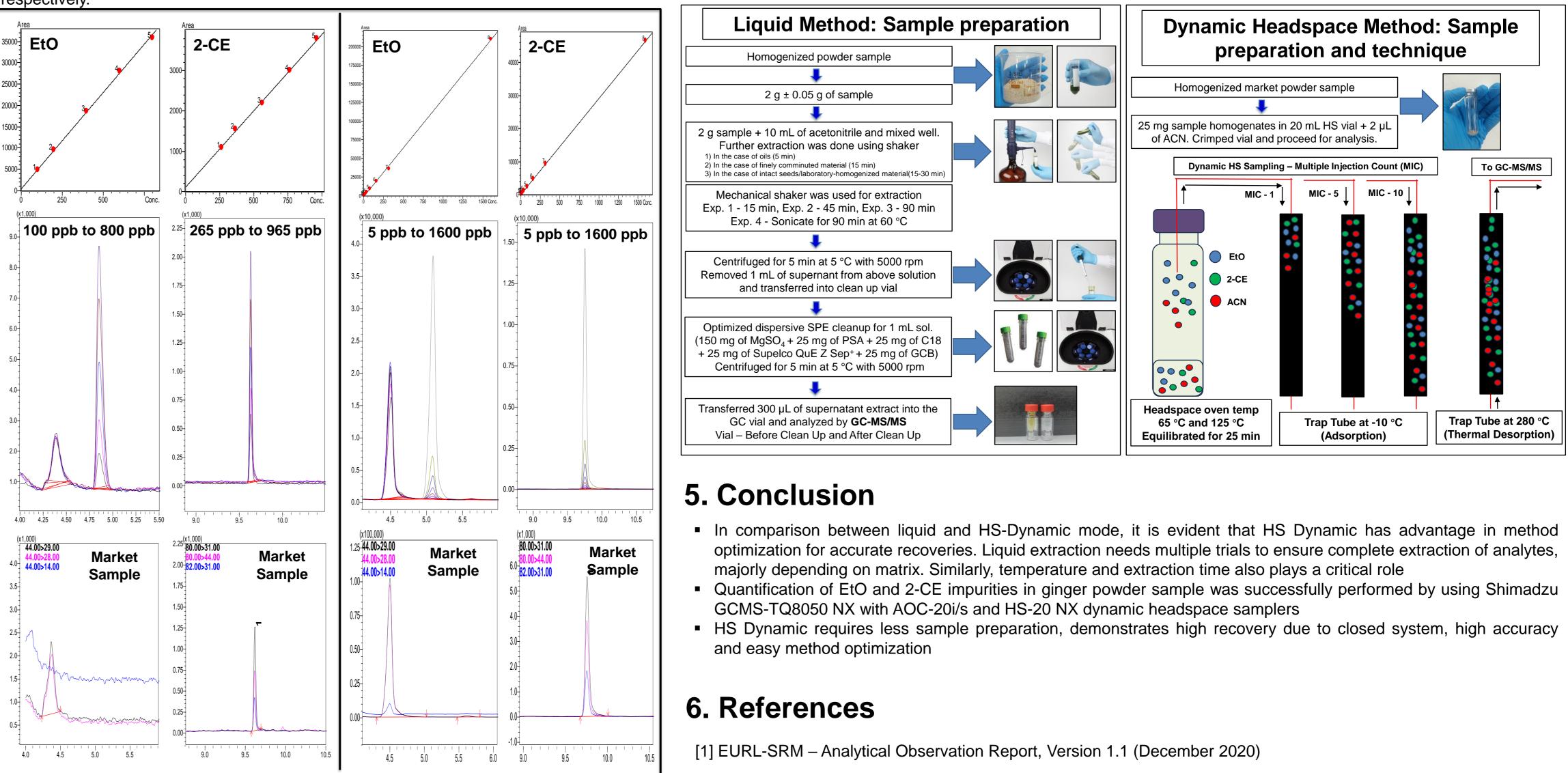
HS - 65°C - Matrix Ma

(5,10,20,40,80,160,320 an

HS - 125°C - Matrix Ma

(5,10,20,40,80,160,320 and

Figure 2 and 3 depict the calibration curve, overlay of linearity standards and representative chromatograms for ginger powder sample for liquid and headspace method, respectively



CE3

44>14

82>31

	Linearity (r ²)		
	EtO	2-CE	
ched	0 00057	0.00644	
800 ppb)	0.99957	0.99641	
atched	0.00017	0.0000	
nd 1600 ppb)	0.99917	0.99999	
latched	0.99984	0.99971	
nd 1600 ppb)	0.99904	0.99971	

Experiment Details		Avg Conc.	ppb
		EtO	
1	Liquid Method (Exp No.1)	-	
2	Liquid Method (Exp No.2)	-	
3	Liquid Method (Exp No.3)	-	
4	Liquid Method (Exp No.4)	-	
1	HS Method (Exp. No.1)	-	
2	HS Method (Exp. No.2)	-	
3	Interlaboratory study	-	

Table 6: Comparison of results

Detailed sample preparation for liquid method and HS method

